## Solution Prep

The solution must be prepared just prior to use. Always add peroxide to the acid.

1. In a clean beaker, add $10 \mathrm{~mL} \mathrm{H} \mathrm{H}_{2} \mathrm{SO}_{4}$
2. Add $2-3 \mathrm{~mL} 30 \% \mathrm{H}_{2} \mathrm{O}_{2}$ to the sulfuric acid mixture
(a) Use a graduated cylinder to measure the hydrogen peroxide. Anything above about 1:1 $\mathrm{H}_{2} \mathrm{O}_{2}: \mathrm{H}_{2} \mathrm{SO}_{4}$ will result in a very violent reaction, potential causing an explosion.
3. The solution will get very hot. Allow it to cool to approximately $80^{\circ} \mathrm{C}$.

## Substrate Cleaning

Piranha solution will quickly remove any organic compounds from the substrate, so only a short time is needed.

1. Use tweezers to place substrate in solution. Be careful not to get the tweezers in the liquid - piranha solution will quickly etch the metal and contaminate the solution.
2. Let sit for about ten minutes. Do not agitate.
3. Pour solution into a separate container for neutralization.
4. Add deionized water to the beaker and remove the substrate.
5. Rinse the substrate in deionized water and blow dry with nitrogen.
