

## BOHR IRON

If each electron contributes 1 Bohr magneton to the magnetization of iron, how many electrons per atom, on average, contribute to the saturation field of 2 Tesla?

$$1 \text{ Bohr magneton} = \frac{e\hbar}{m_e c} = 10^{-23} \text{ amp m}^2 = m$$

$$2 \text{ Tesla} = B = \mu_0 M = \mu_0 m N$$

where  $N$  = density of electrons per  $\text{m}^3$

$$N_{\text{electrons}} = \frac{M}{\mu_0 m} = \frac{2 \text{ Tesla}}{4\pi \times 10^{-7} \times 10^{-23} \text{ amp m}^2/\text{electron}}$$
$$= 1.6 \times 10^{29} \text{ electrons/m}^3$$

density of iron  $\approx 8 \text{ gm/cm}^3$ , atomic weight  $\approx 56$

$$N_{\text{atoms}} \approx 8 \text{ gm/cm}^3 \times 10^6 \text{ cm}^3/\text{m}^3 \times \frac{6 \times 10^{23} \text{ atoms}}{56 \text{ grams}}$$

$$N_{\text{atoms}} \approx 8 \times 10^{28} \text{ atoms/m}^3$$

$$\Rightarrow N_{\text{electrons}} \approx 2 N_{\text{atoms}}, 2 \text{ electrons per atom}$$