Central Forces Homework 10

Due 6/9/17, 4 pm

For every problem, before you start the problem, make a brief statement of the form that a correct solution should have, clearly indicating what quantities you need to solve for. This statement will be graded.

PRACTICE:

- 1. (McIntyre 8.6) Calculate the probability that the electron is measured to be within one Bohr radius of the nucleus for the n = 2 states of hydrogen. Discuss the differences between the results for the l = 0 and l = 1 states.
- 2. Consider the initial state $\frac{1}{\sqrt{2}}(|1,0,0\rangle + |2,1,0\rangle)$. Note, this is **not** an *sp* hybrid orbital such as occurs in chemistry in the study of molecular bonding.
 - (a) If you measure the energy of this state, what possible values could you obtain?
 - (b) What is this state as a function of time?
 - (c) Calculate the expectation value $\langle \hat{L}^2 \rangle$ in this state, as a function of time. Did you expect this answer? Comment.
 - (d) Write the time-dependent state in wave function notation.
 - (e) Calculate the expectation value $\langle \hat{z} \rangle$ as a function of time. Do you expect this answer?
- 3. Complete the attached table for the hydrogen atom.

REQUIRED:

- 4. McIntyre 8.14
- 5. (McIntyre 8.7) Calculate the probability that the electron is measured to be in the classically forbidden region for the n = 2 states of hydrogen. Discuss the differences between the results for the l = -1 and l = 1 states.

Hydrogen Atom

	Ket Representation	Wave Function Representation	Matrix Representation
Hamiltonian			
Eigenvalues of Hamiltonian			
Normalized Eigenstates of Hamiltonian			
Coefficient of the energy eigenstate with quantum numbers n, ℓ, m			
Probability of measuring E_n			

Hydrogen Atom

	Ket Representation	Wave Function Representation	Matrix Representation
Operator for square of the angular momentum			
Eigenvalues of L^2			
Normalized Eigenstates of L^2			
Coefficient of the eigenstates of L^2 with quantum numbers n, ℓ, m			
Probability of measuring $\hbar^2 \ell(\ell+1)$ for the square of the angular momentum			

Hydrogen Atom

	Ket Representation	Wave Function Representation	Matrix Representation
Operator for z- component of angular momentum			
Eigenvalues of L_z			
Normalized Eigenstates of L_z			
Coefficient of m^{th} eigenstates of L_z			
Probability of measuring <i>mħ</i> for <i>z</i> -component of angular momentum			