Chemistry 651

Problem set 3

Due: 23 May 2006

From your text,

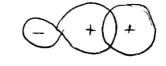
- 1. 7.23
- 2. 7.27
- 3. 7.31
- 4. Use either your text, or the literature to define the bulleted items on page 356 (third edition).
- 5. Write the Hartree Fock equations for the effective one electron HF operator for the $\sigma_g 1s$ and $\pi_g 2p$ orbitals of N_2 using a minimal basis set (1s,2s,2p) for each atom.

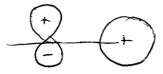
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1, 7.23

c)
$$3dz_a^2 + 3dz_b^2 \Rightarrow \delta g$$
 angular moment in bonding d) $3dx_{0a} + 3dx_{0b} \Rightarrow \delta g$ bonding

2. 7.27





3.7.31

$$| \log^2 | \log^2 | \log^2 20^2 | 30^2 | \pi_u^4 | 1\pi_g^2 \Rightarrow 0_2$$

- a) 4 ret bonding e-
- B 5=1
- c) the bond energy increases it we lose a TTg e od it dicreases it we lose a 30g e -
- d) S = 1/2
- e) Tu >> 2Px