

Energy quantization

Consider a particle moving in 1D in a potential $V(x)$

$$\text{Then } \Psi_E(x, t) = \underbrace{\Psi_E(x)}_{\substack{\uparrow \\ \text{energy eigenfunctions}}} e^{-\frac{i}{\hbar} Et}$$

\Rightarrow

$$H \Psi_E(x) \equiv -\frac{\hbar^2}{2m} \frac{d^2 \Psi_E(x)}{dx^2} + V(x) \Psi_E(x) = E \Psi_E(x)$$

\Downarrow

$$\underbrace{\frac{d^2 \Psi_E(x)}{dx^2}}_{\substack{\Downarrow \\ \text{}}}= \frac{2m}{\hbar^2} [V(x) - E] \Psi_E(x)$$

Let's analyze this equation:

2nd order diff. eq. \Rightarrow has two linearly independent solutions
for any E

If $V(x)$ is finite $\Rightarrow \Psi''_E(x)$ is finite $\Rightarrow \Psi'_E(x)$ and $\Psi_E(x)$ are continuous

Out of all mathematical solutions, we have (2)
 To choose those that are physical $\Rightarrow \Psi(x)$ must be
 finite and single-valued everywhere.
 need to eliminate unphysical solutions

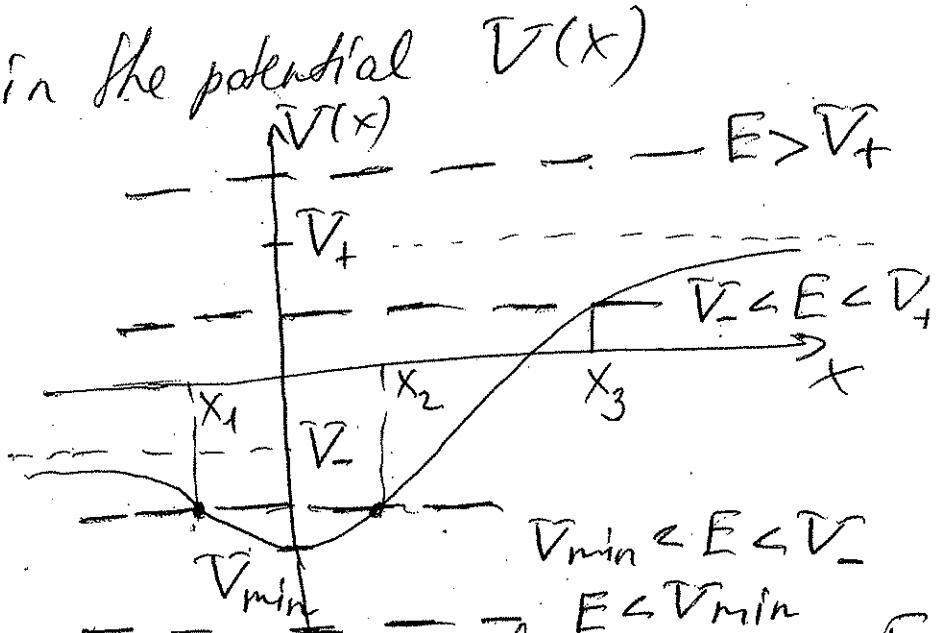
Example

Consider a particle in the potential $V(x)$

$V \rightarrow V_+$ at $x \rightarrow +\infty$

$V \rightarrow V_-$ at $x \rightarrow -\infty$

$V = V_{\min}$ at $x=0$



Four cases can be distinguished, depending on the value of energy E

Case 1 $E < V_{\min}$

$$\frac{d^2\Psi_E(x)}{dx^2} = \frac{2m}{\hbar^2} (V(x) - E) \Psi_E(x) \Rightarrow \Psi''_E \text{ and } \Psi_E > 0 \text{ at all } x \text{'s}$$

are always of the same sign!

if $\Psi_E > 0 \Rightarrow \Psi''_E > 0$

$$\Psi_E < 0 \Rightarrow \Psi''_E < 0$$

Recall from functional analysis \Rightarrow a sign of the 2nd

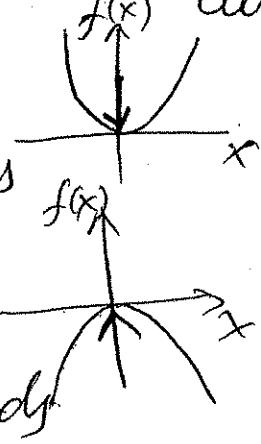
derivative shows (determines) the sign of the ⁽³⁾
 $f(x)$ curvature

Example: $f = x^2$; $f'' = 2 > 0$

$f = -x^2$; $f'' = -2 < 0$

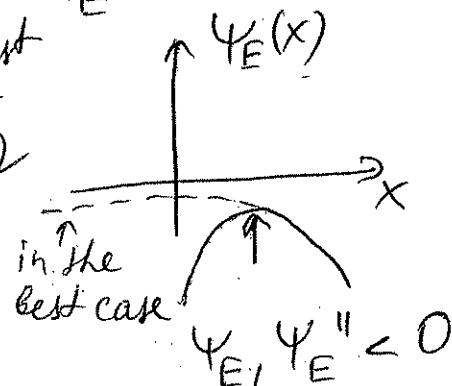
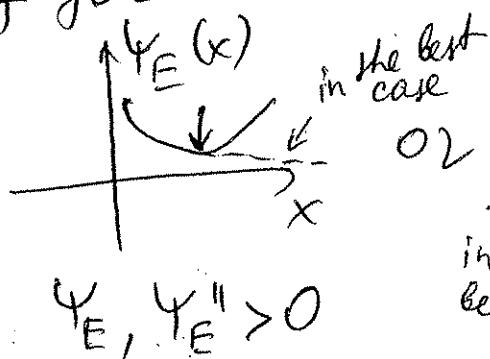
Concave upwards

Concave downwards



Back to our Case 1

If for all x 's ψ_E'' is either > 0 or $< 0 \Rightarrow$



\Rightarrow In both cases
 $\psi_E(x)$ blows up
 either at $x \rightarrow \infty$,
 or at $x \rightarrow -\infty$
 or both!

Also note that

no physical solution
in this case !

if $V(x) - E > 0$

\Rightarrow kinetic energy ↑

↑
 $T + V$ is negative at all x 's !

(classically - no motion is possible)

Case 2

$V_{min} < E < V_-$

At $x = x_1$ and $x = x_2 \Rightarrow E = V(x) \Leftarrow$ These are the
 classical limits of the motion of a classical particle of energy E ← turning points

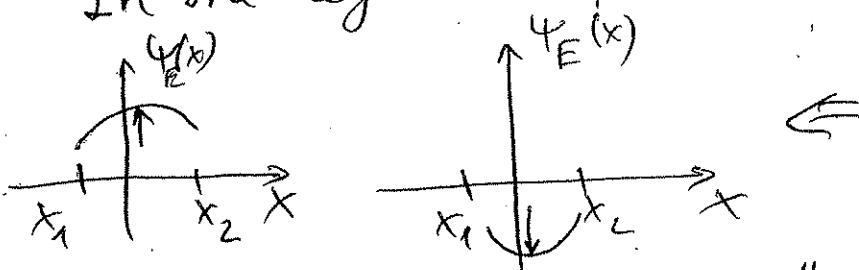
So, classically, the motion is confined to (7)

$x_1 \leq x \leq x_2$ (at $x < x_1$ and $x > x_2$ the kinetic energy becomes negative)

What about quantum-mechanically?

Consider three regions: $x < x_1$, $x_1 \leq x \leq x_2$, $x > x_2$

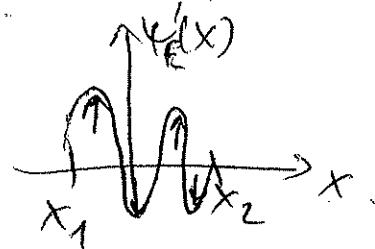
In the region $x_1 \leq x \leq x_2 \Rightarrow V(x) - E < 0 \Rightarrow$



$$\Psi_E > 0, \Psi''_E < 0 \quad \Psi_E < 0, \Psi''_E > 0$$

Ψ''_E is of opposite sign
to Ψ_E .

Oscillatory behavior (Once Ψ_E reaches 0, its curvature flips) \Rightarrow
(two linearly independent oscillatory solutions)



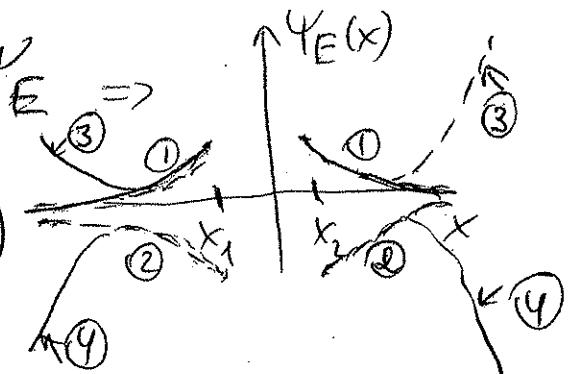
In the regions $[x > x_2]$

Ψ''_E is of the same sign as Ψ_E

solutions like ① or ②

are physical ($\Psi_E \rightarrow 0$ at $x \rightarrow \pm\infty$)

③ & ④ are unphysical



Example: $\Psi''_E - k^2 \Psi_E = 0 \Rightarrow \Psi_E(x) = C_1 e^{kx} + C_2 e^{-kx}$

If consider $x > x_2 \Rightarrow$ have to have $C_1 = 0$ { to keep $\Psi_E(x)$ finite at $x \rightarrow \pm\infty$ }
 $x < x_1 \Rightarrow C_2 = 0$

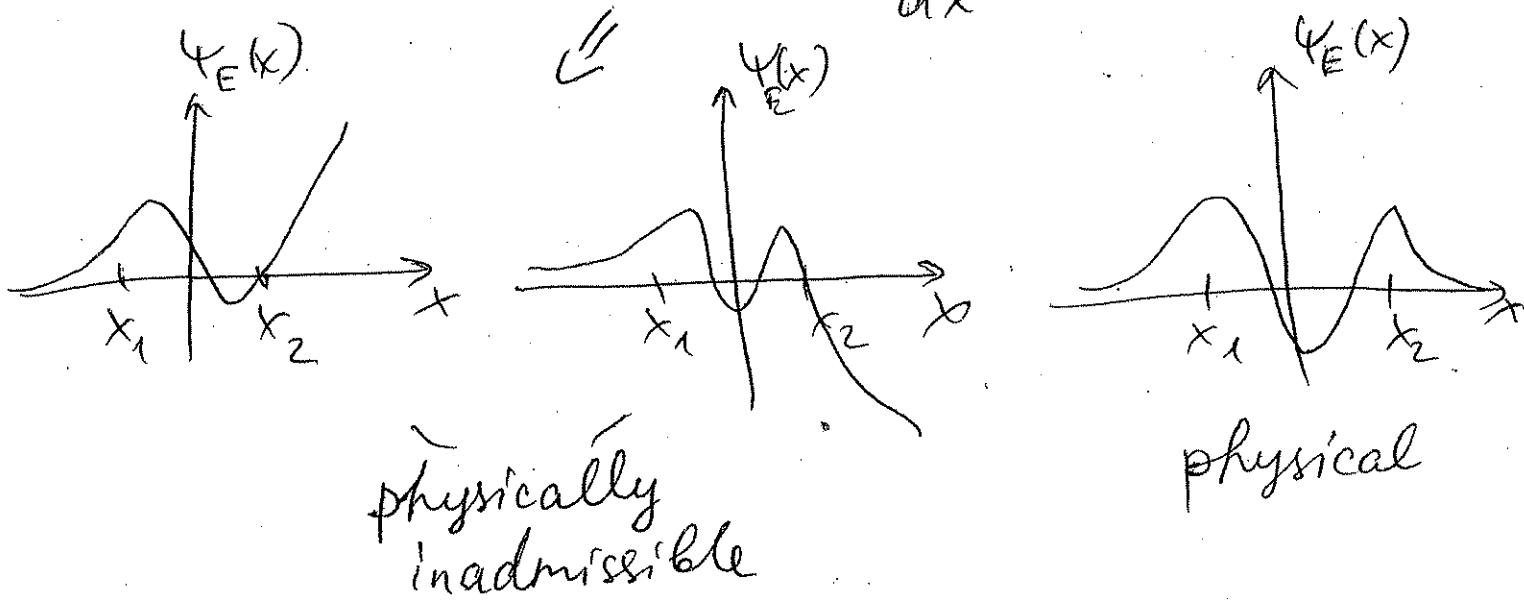
So, when joining the three regions, we need (5) to ensure that: 1) $\Psi_E(x)$ is finite at $x \rightarrow \pm\infty$

2) $\Psi_E(x)$ is continuous

3) if $V(x)$ is finite everywhere,

Note:
if $V(x) = C\delta(x-x_0)$ \Rightarrow
then it's not the case

then $\frac{d\Psi_E(x)}{dx}$ is continuous



Since $\Psi_E(x) \rightarrow 0$ (and the probability density beyond $x_1 \leq x \leq x_2$ $| \Psi_E(x) |^2$) \Rightarrow these eigenfunctions

represent bound states. The spectrum of the bound states is discrete \Rightarrow quantization of energy.

The number of the discrete energy levels depends on $V(x)$ \Rightarrow can be finite or infinite.

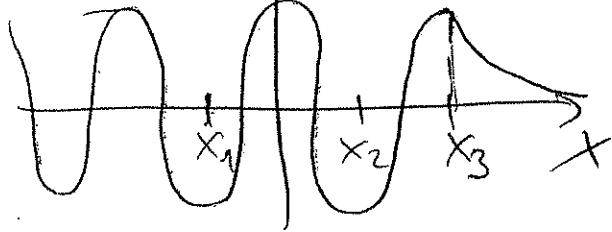
In 1D the bound-state (discrete) energy eigenvalues are non-degenerate (not true for 2D or 3D)

Case 3 $V_- < E < V_+$ (6)

At this energy interval, there is only one classical turning point at $x = x_3$ (no classical motion is allowed at $x > x_3$). Classically the particle is the particle moving unbound (since it can move in to the right would be an infinite region of the x-axis) reflected

At $x < x_3 \Rightarrow$ get oscillatory solutions $\Psi_E(x)$

At $x > x_3 \Rightarrow$ get one $\Psi_E(x) \rightarrow 0$ as $x \rightarrow \infty$ and $\Psi_E(x)$ another one $\Psi_E(x) \rightarrow \infty$ as $x \rightarrow \infty$ \downarrow unphysical



In this case, "smooth matching" of $\Psi_E(x)$ and $\Psi'_E(x)$ at $x = x_3$ can be made at all energies

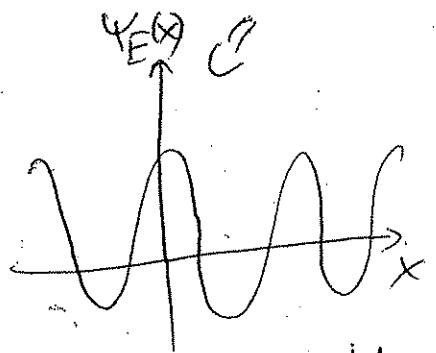
$V_- < E < V_+ \Rightarrow$ allowed energies form a continuum and are non-degenerate

Wave functions $\Psi_E(x)$ which are non-zero and finite at infinity (in this case at $-\infty$) correspond to unbound (or scattering) states.

Case 4 $E > V_+$ (7)

$\rightarrow E - V(x) > 0$ for all x 's \Rightarrow

$\Psi_E''(x)$ is opposite in sign to $\Psi_E(x)$ for all $x \Rightarrow$
two oscillatory solutions which are physically
admissible for every energy. So at $E > V_+$,
the energy eigenvalues form a continuum and are
doubly degenerate (because of the two independent
states of motion \Rightarrow). The eigenfunctions $\Psi_E(x)$
describe unbound states

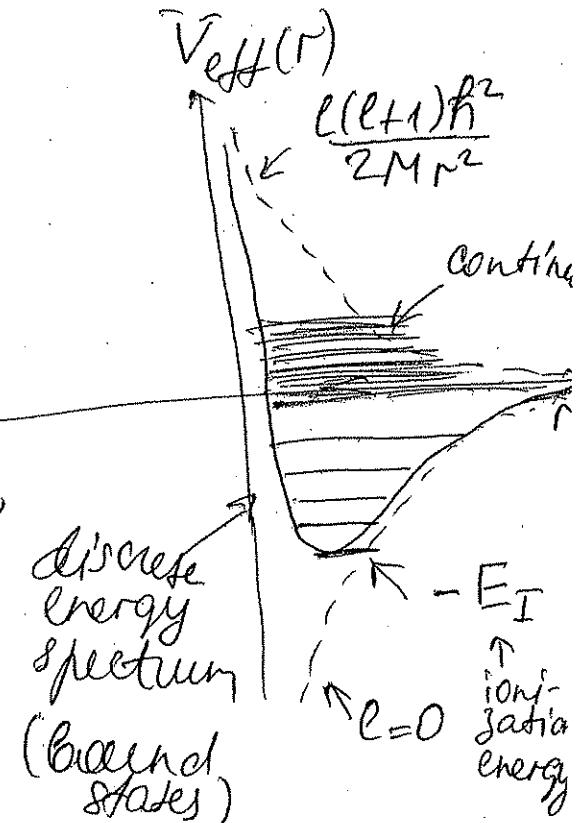


Example Hydrogen atom (H)

$$V_{\text{eff}}(r) = V(r) + \frac{l(l+1)\hbar^2}{2Mr^2}$$

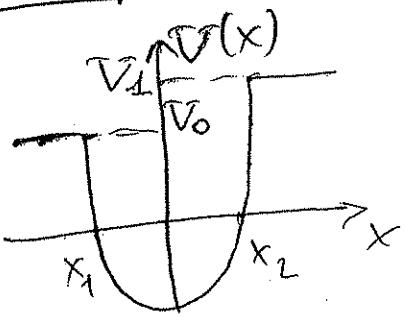
\uparrow Coulomb $\frac{l(l+1)\hbar^2}{2Mr^2}$ centrifugal
 $\frac{-e^2}{r}$ potential

Normally we care about bound states,
since unbound states here represent
the electron after ionisation of H,
i.e. free electron \Leftarrow this is of interest
in particle collision/scattering though

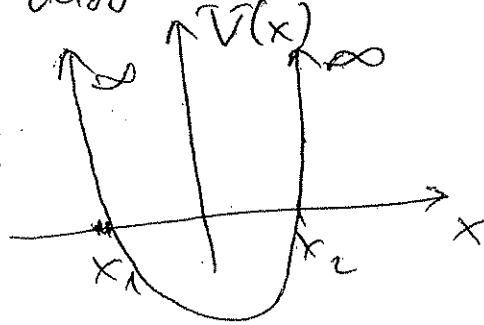


Until now we've considered finite $V(x)$. (8)
 What if $V = \infty$? \Rightarrow because a particle cannot have an infinite energy, it cannot penetrate in a region where $V = \infty \Rightarrow \Psi_E = 0$ in that region.

Example How are the energy spectra of these two systems different?



two systems different?



①

$E < V_0 \Rightarrow$ discrete spectrum

$E > V_0 \Rightarrow$ continuum

②

For all E 's \Rightarrow discrete allowed energy levels determined by boundary conditions $\Psi(x_1) = \Psi(x_2) = 0$